Atomic Mass Activity						
The purpose of this activity is to explore atomic mass. You will use the following representation for sub-atomic particles:						
• Protons: dark-colored marbles						
• Neutrons: light-colored marbles						
• Electrons: paper circles						
1. Find the average mass of your marbles (protons & neutrons) by weighing all of them to the nearest 0.1 g on a balance, and dividing by the number of marbles.						
Total mass: g						
What is the name of the unit of measurement that your average mass of a marble represen						
2. Find the element carbon (C) on the periodic table. Its atomic number is 6, which means that it has six protons. Place 6 protons (dark-colored marbles) in your cup.						
3. Add six neutrons (light-colored marbles) and six electrons (paper circles) and measure mass on your balance.						
Mass of carbon with 6 protons, 6 neutrons, & 6 electrons: g						
4. Divide the mass by your average mass from step $#1$.						
"Atomic mass" of carbon with 6 protons, 6 neutrons, & 6 electrons: amu						
5. Write the chemical symbol for carbon with 6 protons, 6 neutrons, & 6 electrons:						
6. Add one more neutron. You now have carbon with 6 protons, 7 neutrons, and 6 electron						
Actual Mass: g Atomic mass: amu Chem. symbol:						
7. Add four more electrons.						
Actual Mass: g Atomic mass: amu Chem. symbol:						

Name: _____

Block: _____

8.	Repeat this process for each of the following elements:					
	(a)	Nitrogen with 7 protons, 7	7 neutrons and 7 electron	ns		
		Actual mass: g	Atomic mass:	amu	Chem. symbol:	
	(b)	Nitrogen with 7 protons, 7 neutrons and 10 electrons				
		Actual mass: g	Atomic mass:	amu	Chem. symbol:	
	(c)	Nitrogen with 7 protons, 8 neutrons, and 7 electrons				
		Actual mass: g	Atomic mass:	amu	Chem. symbol:	
	(d)	Nitrogen with 7 protons, 9 neutrons, and 10 electrons				
		Actual mass: g	Atomic mass:	amu	Chem. symbol:	
	(e)	Lithium with 3 protons, 3 neutrons, and 3 electrons				
		Actual mass: g	Atomic mass:	amu	Chem. symbol:	
	(f)	(f) Lithium with 3 protons, 4 neutrons, and 2 electrons				
		Actual mass: g	Atomic mass:	amu	Chem. symbol:	
	(g)	Hydrogen with 1 proton, no neutrons, and 1 electron				
		Actual mass: g	Atomic mass:	amu	Chem. symbol:	
	(h)	Hydrogen with 1 proton, 2 neutrons, and 1 electron				
		Actual mass: g	Atomic mass:	amu	Chem. symbol:	
	(i)	helium with 2 protons, 1 neutron, and 2 electrons				
		Actual mass: g	Atomic mass:	amu	Chem. symbol:	
	(j)	helium with 2 protons, 2 neutrons, and 2 electrons				
		Actual mass: g	Atomic mass:	amu	Chem. symbol:	