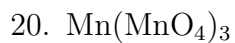
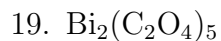
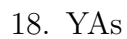
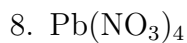
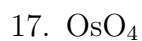
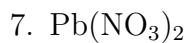
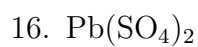
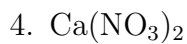


Name: _____

Block: _____

Naming Ionic Compounds #2

Give the stock name for each of the following compounds.



Give the formula for each of the following compounds.

21. potassium hydroxide

31. lithium bromide

22. strontium iodide

32. lead (IV) sulfate

23. iron (II) nitride

33. silver nitrate

24. zinc chloride

34. ammonium sulfate

25. uranyl phosphate

35. copper (I) oxide

26. cobalt (II) chloride

36. copper (II) oxide

27. titanium (IV) oxide

37. mercury (I) chloride

28. gallium nitrate

38. mercury (II) phosphate

29. barium hydroxide

39. ytterbium (II) sulfide

30. tantalum nitride

40. chromium (VI) nitride